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# Mulliken's Electronegativity Formulas

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# List of 9 Mulliken's Electronegativity Formulas

## Mulliken's Electronegativity

### 1) Covalent Radius given Mulliken's Electronegativity

$$\text{fx } r_{\text{covalent}} = \sqrt{\frac{0.359 \cdot Z}{(0.336 \cdot X_M) - 0.2 - 0.744}}$$

Open Calculator 

$$\text{ex } 1.17979\text{\AA} = \sqrt{\frac{0.359 \cdot 25}{(0.336 \cdot 22J) - 0.2 - 0.744}}$$

### 2) Effective Nuclear Charge given Mulliken's Electronegativity

$$\text{fx } Z = \frac{((0.336 \cdot X_M) - 0.2 - 0.744) \cdot (r_{\text{covalent}}^2)}{0.359}$$

Open Calculator 

$$\text{ex } 25.0089 = \frac{((0.336 \cdot 22J) - 0.2 - 0.744) \cdot ((1.18\text{\AA})^2)}{0.359}$$

### 3) Electron Affinity of element using Mulliken's Electronegativity

$$\text{fx } \text{E.A} = (2 \cdot X_M) - \text{IE}$$

Open Calculator 

$$\text{ex } 16.8J = (2 \cdot 22J) - 27.2J$$



4) Ionization Energy of element using Mulliken's Electronegativity 

$$\text{fx } IE = (2 \cdot X_M) - E.A$$

[Open Calculator !\[\]\(cbe80b694ebd74fcfe136a095b608235\_img.jpg\)](#)

$$\text{ex } 26.9\text{J} = (2 \cdot 22\text{J}) - 17.1\text{J}$$

5) Mulliken's Electronegativity from Allred Rochow's Electronegativity 

$$\text{fx } X_M = \frac{X_{A.R} + 0.744 + 0.2}{0.336}$$

[Open Calculator !\[\]\(3e2231b1ad3ca8da8658228c00dd08e0\_img.jpg\)](#)

$$\text{ex } 22.15476\text{J} = \frac{6.5\text{J} + 0.744 + 0.2}{0.336}$$

6) Mulliken's Electronegativity from Pauling's Electronegativity 

$$\text{fx } X_M = \frac{X_P + 0.2}{0.336}$$

[Open Calculator !\[\]\(0d5ec72f61334709c3fc9450209b754f\_img.jpg\)](#)

$$\text{ex } 22.14286\text{J} = \frac{7.24\text{J} + 0.2}{0.336}$$

7) Mulliken's Electronegativity given Bond Energies 

$$\text{fx } X_M = \frac{\sqrt{E_{(A-B)}} - \sqrt{E_{A-A} \cdot E_{B-B}} + 0.2}{0.336}$$

[Open Calculator !\[\]\(b64b40baaee5acddc1eab8538ba84754\_img.jpg\)](#)

$$\text{ex } 22.1047\text{J} = \frac{\sqrt{75.47\text{J}} - \sqrt{20\text{J} \cdot 27\text{J}} + 0.2}{0.336}$$



## 8) Mulliken's Electronegativity given Effective Nuclear Charge and Covalent Radius

$$\text{fx } X_M = \frac{\left( \frac{0.359 \cdot Z}{r_{\text{covalent}}^2} \right) + 0.744 + 0.2}{0.336}$$

[Open Calculator !\[\]\(e78f798d4ea5c530c9db49e7d26e6b95\_img.jpg\)](#)

$$\text{ex } 21.99317J = \frac{\left( \frac{0.359 \cdot 25}{(1.18\text{\AA})^2} \right) + 0.744 + 0.2}{0.336}$$

## 9) Mulliken's Electronegativity of Element

$$\text{fx } X_M = 0.5 \cdot (\text{IE} + \text{E.A})$$

[Open Calculator !\[\]\(05be7c7a8995decd503647c99211f7c2\_img.jpg\)](#)

$$\text{ex } 22.15J = 0.5 \cdot (27.2J + 17.1J)$$





## Variables Used

- $E_{(A-B)}$  Actual Bond Energy given Electronegativity (Joule)
- $E_{A-A}$  Bond Energy of  $A_2$  Molecule (Joule)
- $E_{B-B}$  Bond Energy of  $B_2$  Molecule (Joule)
- $E.A$  Electron Affinity (Joule)
- $IE$  Ionization Energy (Joule)
- $r_{\text{covalent}}$  Covalent Radius (Angstrom)
- $X_{A.R}$  Allred-Rochow's Electronegativity (Joule)
- $X_M$  Mulliken's Electronegativity (Joule)
- $X_P$  Pauling's Electronegativity (Joule)
- $Z$  Effective Nuclear Charge






## Constants, Functions, Measurements used

- **Function:** **sqrt**, sqrt(Number)  
*Square root function*
- **Measurement:** **Length** in Angstrom (A)  
*Length Unit Conversion* 
- **Measurement:** **Energy** in Joule (J)  
*Energy Unit Conversion* 



## Check other formula lists

- [Allred Rochow's Electronegativity Formulas](#) 
- [Pauling's Electronegativity Formulas](#) 
- [Mulliken's Electronegativity Formulas](#) 

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