



Important Formulas in Mass Transfer Coefficient, Driving Force and Theories

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List of 29 Important Formulas in Mass Transfer Coefficient, Driving Force and Theories

Important Formulas in Mass Transfer Coefficient, Driving Force and Theories C

1) Average Mass Transfer Coefficient by Penetration Theory

fx
$$k_{
m L\,(Avg)} = 2 \cdot \sqrt{rac{{
m D}_{
m AB}}{\pi \cdot {
m t}_{
m c}}}$$

ex
$$0.028465 \mathrm{m/s} = 2 \cdot \sqrt{rac{0.007 \mathrm{m^2/s}}{\pi \cdot 11 \mathrm{s}}}$$

2) Average Sherwood Number of Combined Laminar and Turbulent Flow

$$\mathbf{x} \left[\mathrm{N}_{\mathrm{sh}} = \left(\left(0.037 \cdot \left(\mathrm{Re}^{0.8}
ight)
ight) - 871
ight) \cdot \left(\mathrm{Sc}^{0.333}
ight)
ight)$$

Open Calculator

Open Calculator

$$\mathbf{x}$$
 1074.78 = $\left(\left(0.037\cdot\left((500000)^{0.8}
ight)
ight) - 871
ight)\cdot\left((12)^{0.333}
ight)$

3) Average Sherwood Number of Flat Plate Turbulent Flow

fx
$$\mathrm{N_{sh}}=0.037\cdot\left(\mathrm{Re}^{0.8}
ight)$$

x
$$1340.842 = 0.037 \cdot \left(\left(500000
ight)^{0.8}
ight)$$



e

e

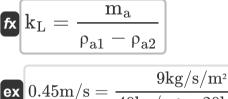


4) Average Sherwood Number of Internal Turbulent Flow 🕑

fx
$$\mathrm{N_{sh}} = 0.023 \cdot (\mathrm{Re}^{0.83}) \cdot (\mathrm{Sc}^{0.44})$$

$$\fbox{\textbf{ex}}3687.336 = 0.023 \cdot \left((500000)^{0.83} \right) \cdot \left((12)^{0.44} \right)$$

5) Convective Mass Transfer Coefficient 🕑



$$0.45 \text{m/s} = \frac{9 \text{kg/s/m}^2}{40 \text{kg/m}^3 - 20 \text{kg/m}^3}$$

6) Convective Mass Transfer Coefficient for Simultaneous Heat and Mass Transfer

$$\begin{split} & \textbf{fx} \ \textbf{k}_L = \frac{h_t}{Q_s \cdot \rho_L \cdot \left(L_e^{0.67}\right)} & \textbf{Open Calculator C} \\ & \textbf{ex} \ \textbf{4E}^{-5m/s} = \frac{13.2W/m^{2*}K}{120J/(kg^*K) \cdot 1000kg/m^3 \cdot \left((4.5)^{0.67}\right)} \end{split}$$



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7) Convective Mass Transfer Coefficient of Flat Plate in Combined Laminar Turbulent Flow

$$\mathbf{k}_{\mathrm{L}} = rac{0.0286 \cdot \mathrm{u}_{\infty}}{\left(\mathrm{Re}^{0.2}
ight) \cdot \left(\mathrm{Sc}^{0.67}
ight)}$$

Open Calculator 🖸

$$\mathbf{x} 0.004118 \text{m/s} = \frac{0.0286 \cdot 10.5 \text{m/s}}{\left((500000)^{0.2} \right) \cdot \left((12)^{0.67} \right)}$$

8) Convective Mass Transfer Coefficient of Flat Plate Laminar Flow using Drag Coefficient

fx
$$\mathbf{k}_{\mathrm{L}} = rac{\mathrm{C}_{\mathrm{D}} \cdot \mathbf{u}_{\infty}}{2 \cdot \left(\mathrm{Sc}^{0.67}
ight)}$$
 ex $29.80088 \mathrm{m/s} = rac{30 \cdot 10.5 \mathrm{m/s}}{2 \cdot \left(\left(12\right)^{0.67}
ight)}$

Open Calculator 🗗

9) Convective Mass Transfer Coefficient of Flat Plate Laminar Flow using Friction Factor

fx
$$\mathbf{k}_{\mathrm{L}} = rac{\mathbf{f} \cdot \mathbf{u}_{\infty}}{8 \cdot \left(\mathbf{Sc}^{0.67}\right)}$$

ex $0.156455 \mathrm{m/s} = rac{0.63 \cdot 10.5 \mathrm{m/s}}{8 \cdot \left((12)^{0.67}\right)}$

Open Calculator



10) Convective Mass Transfer Coefficient of Flat Plate Laminar Flow using Reynolds Number

fx
$$\mathbf{k}_{\mathrm{L}} = rac{\mathrm{u}_{\infty} \cdot 0.322}{\left(\mathrm{Re}^{0.5}
ight) \cdot \left(\mathrm{Sc}^{0.67}
ight)}$$

Open Calculator 🕑

Open Calculator

ex
$$0.000905 \mathrm{m/s} = rac{10.5 \mathrm{m/s} \cdot 0.322}{\left(\left(500000
ight)^{0.5}
ight) \cdot \left(\left(12
ight)^{0.67}
ight)}$$

11) Convective Mass Transfer Coefficient through Liquid Gas Interface

fx
$$\mathbf{k}_{\mathrm{L}} = rac{\mathrm{m}_{1} \cdot \mathrm{m}_{2} \cdot \mathrm{H}}{(\mathrm{m}_{1} \cdot \mathrm{H}) + (\mathrm{m}_{2})}$$

ex
$$0.004767 \mathrm{m/s} = rac{0.3 \mathrm{m/s} \cdot 0.7 \mathrm{m/s} \cdot 0.016}{(0.3 \mathrm{m/s} \cdot 0.016) + (0.7 \mathrm{m/s})}$$

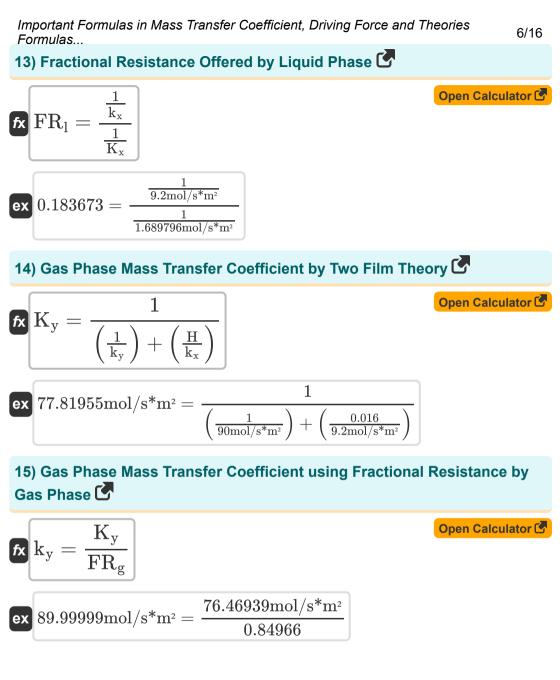
12) Fractional Resistance Offered by Gas Phase 🕑

$$fx FR_g = \frac{\frac{1}{k_y}}{\frac{1}{K_y}}$$

$$ex 0.84966 = \frac{\frac{1}{90 \text{mol/s}^*\text{m}^2}}{\frac{1}{76.46939 \text{mol/s}^*\text{m}^2}}$$











16) Heat Transfer Coefficient for Simultaneous Heat and Mass Transfer 🕑

fx
$$\mathbf{h_t} = \mathbf{k_L} \cdot \mathbf{
ho_L} \cdot \mathbf{Q_s} \cdot \left(\mathbf{L_e^{0.67}}
ight)$$

ex

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 $\Big| 1479.266 \mathrm{W/m^{2}}^{*}\mathrm{K} = 4.5\mathrm{e}{-}3\mathrm{m/s} \cdot 1000 \mathrm{kg/m^{3}} \cdot 120 \mathrm{J/(kg^{*}\mathrm{K})} \cdot \left((4.5)^{0.67}
ight)$

17) Liquid Phase Mass Transfer Coefficient by Two Film Theory 🕑



ex
$$1.245113 \text{mol/s}^*\text{m}^2 = \frac{1}{\left(\frac{1}{90 \text{mol/s}^*\text{m}^2 \cdot 0.016}\right) + \left(\frac{1}{9.2 \text{mol/s}^*\text{m}^2}\right)}$$

18) Liquid Phase Mass Transfer Coefficient using Fractional Resistance by Liquid Phase

$$\begin{aligned} & \textbf{K}_{x} = \frac{K_{x}}{FR_{l}} \\ & \textbf{K}_{x} = \frac{K_{x}}{FR_{l}} \\ & \textbf{K}_{x} = \frac{K_{x}}{FR_{l}} \\ & \textbf{K}_{x} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{K}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.183673} \\ & \textbf{M}_{x} = 0.200024 \text{mol/s}^{*}\text{m}^{2} = \frac{1.689796 \text{mol/s}^{*}\text{m}^{2}}{0.18367} \\$$

Important Formulas in Mass Transfer Coefficient, Driving Force and Theories8/16**20) Local Sherwood Number for Flat Plate in Turbulent Flow**Image: Comparison of the second sec

Open Calculator

fx
$$\mathrm{L_{sh}}=0.0296\cdot\left(\mathrm{Re}_{1}^{0.8}
ight)\cdot\left(\mathrm{Sc}^{0.333}
ight)$$

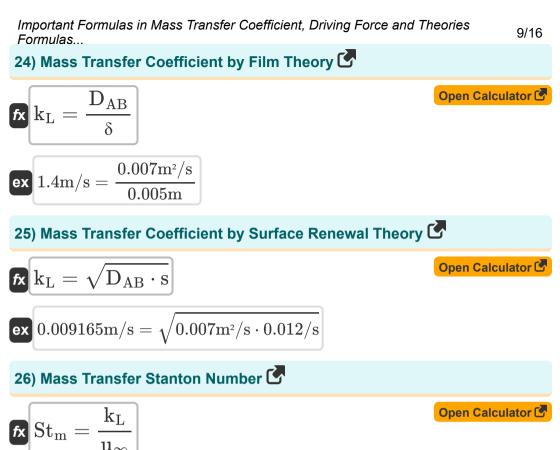
$$\mathbf{x} \left[0.041971 = 0.0296 \cdot \left((0.55)^{0.8} \right) \cdot \left((12)^{0.333} \right) \right]$$

fx
$$C_{bm} = \frac{C_{b2} - C_{b1}}{\ln\left(\frac{C_{b2}}{C_{b1}}\right)}$$

ex $12.33152 \text{mol/L} = \frac{10 \text{mol/L} - 15 \text{mol/L}}{\ln\left(\frac{10 \text{mol/L}}{15 \text{mol/L}}\right)}$

22) Logarithmic Mean Partial Pressure Difference 🕑

$$\begin{aligned} & \mathsf{P}_{bm} = \frac{\mathsf{P}_{b2} - \mathsf{P}_{b1}}{\ln\left(\frac{\mathsf{P}_{b2}}{\mathsf{P}_{b1}}\right)} \\ & \bullet \\ & \mathsf{P}_{bm} = \frac{10500\mathsf{Pa} - 8700\mathsf{Pa}}{\ln\left(\frac{10500\mathsf{Pa}}{8700\mathsf{Pa}}\right)} \\ & \bullet \\ & \bullet \\ & \mathsf{P}_{bm} = \frac{10500\mathsf{Pa} - 8700\mathsf{Pa}}{\ln\left(\frac{10500\mathsf{Pa}}{8700\mathsf{Pa}}\right)} \\ & \bullet \\ & \mathsf{S}_{star} = \mathsf{S}_{star} \cdot \mathsf{S}_{star} \\ & \mathsf{S}_{star} = \mathsf{S}_{bt} \cdot \mathsf{S}_{star} \\ & \mathsf{S}_{star} = \mathsf{S}_{bt} \cdot \mathsf{S}_{star} \\ & \bullet \\ & \mathsf{S}_{star} = \mathsf{S}_{bt} \cdot \mathsf{S}_{star} \\ & \bullet \\ & \bullet$$



ex
$$0.000429 = rac{4.5 ext{e-3m/s}}{10.5 ext{m/s}}$$

27) Overall Gas Phase Mass Transfer Coefficient using Fractional Resistance by Gas Phase

fx
$$\mathrm{K_y} = \mathrm{k_y} \cdot \mathrm{FR_g}$$

$$= 76.4694 \text{mol/s}^*\text{m}^2 = 90 \text{mol/s}^*\text{m}^2 \cdot 0.84966$$



Open Calculator

28) Overall Liquid Phase Mass Transfer Coefficient using Fractional Resistance by Liquid Phase

$$\label{eq:Kx} \begin{array}{l} \text{fx} \ K_x = k_x \cdot FR_1 \\ \text{ex} \ 1.689792 \text{mol}/\text{s}^*\text{m}^2 = 9.2 \text{mol}/\text{s}^*\text{m}^2 \cdot 0.183673 \\ \text{ex} \ 1.689792 \text{mol}/\text{s}^*\text{m}^2 = 9.2 \text{mol}/\text{s}^*\text{m}^2 \cdot 0.183673 \\ \text{29) Sherwood Number for Flat Plate in Laminar Flow } \\ \text{fx} \ N_{sh} = 0.664 \cdot \left(\text{Re}^{0.5}\right) \cdot \left(\text{Sc}^{0.333}\right) \\ \text{open Calculator } \\ \text{ex} \ 1074.04 = 0.664 \cdot \left((500000)^{0.5}\right) \cdot \left((12)^{0.333}\right) \\ \end{array}$$





Variables Used

- **C**_{b1} Concentration of Component B in Mixture 1 (*Mole per Liter*)
- Cb2 Concentration of Component B in Mixture 2 (Mole per Liter)
- C_{bm} Logarithmic Mean of Concentration Difference (*Mole per Liter*)
- C_D Drag Coefficient
- **D**_{AB} Diffusion Coefficient (DAB) (Square Meter Per Second)
- **f** Friction Factor
- FR_a Fractional Resistance Offered by Gas Phase
- FRI Fractional Resistance Offered by Liquid Phase
- H Henry's Constant
- **h**_t Heat Transfer Coefficient (Watt per Square Meter per Kelvin)
- k_{L (Avg)} Average Convective Mass Transfer Coefficient (Meter per Second)
- **k**_L Convective Mass Transfer Coefficient (Meter per Second)
- **k**_L Convective Mass Transfer Coefficient (Meter per Second)
- k_x Liquid Phase Mass Transfer Coefficient (Mole per Second Square Meter)
- K_X Overall Liquid Phase Mass Transfer Coefficient (Mole per Second Square Meter)
- **k**_v Gas Phase Mass Transfer Coefficient (Mole per Second Square Meter)
- K_y Overall Gas Phase Mass Transfer Coefficient (Mole per Second Square Meter)
- Lewis Number





- L_{sh} Local Sherwood Number
- m₁ Mass Transfer Coefficient of Medium 1 (Meter per Second)
- m2 Mass Transfer Coefficient of Medium 2 (Meter per Second)
- m_a Mass Flux of Diffusion Component A (Kilogram per Second per Square Meter)
- N_{sh} Average Sherwood Number
- **P**_{b1} Partial Pressure of Component B in 1 (Pascal)
- Pb2 Partial Pressure of Component B in 2 (Pascal)
- Pbm Logarithmic Mean Partial Pressure Difference (Pascal)
- **Q**_S Specific Heat (Joule per Kilogram per K)
- Re Reynolds Number
- Rel Local Reynolds Number
- S Surface Renewal Rate (1 Per Second)
- Sc Schmidt Number
- St_m Mass Transfer Stanton Number
- **t**_c Average Contact Time (Second)
- U_∞ Free Stream Velocity (Meter per Second)
- δ Film Thickness (Meter)
- δ_{mx} Mass Transfer Boundary Layer Thickness at x
- **ρ**_{a1} Mass Concentration of Component A in Mixture 1 (Kilogram per Cubic Meter)
- **ρ_{a2}** Mass Concentration of Component A in Mixture 2 (Kilogram per Cubic Meter)
- ρ_L Density of Liquid (Kilogram per Cubic Meter)





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• δ_{hx} Hydrodynamic Boundary Layer Thickness (*Meter*)



Constants, Functions, Measurements used

- Constant: pi, 3.14159265358979323846264338327950288 Archimedes' constant
- Function: In, In(Number) The natural logarithm, also known as the logarithm to the base e, is the inverse function of the natural exponential function.
- Function: sqrt, sqrt(Number) A square root function is a function that takes a non-negative number as an input and returns the square root of the given input number.
- Measurement: Length in Meter (m) Length Unit Conversion
- Measurement: Time in Second (s) Time Unit Conversion
- Measurement: Pressure in Pascal (Pa) Pressure Unit Conversion
- Measurement: Speed in Meter per Second (m/s)
 Speed Unit Conversion
- Measurement: Specific Heat Capacity in Joule per Kilogram per K
 (J/(kg*K))

Specific Heat Capacity Unit Conversion 🗹

 Measurement: Heat Transfer Coefficient in Watt per Square Meter per Kelvin (W/m^{2*}K)

Heat Transfer Coefficient Unit Conversion 🖒

- Measurement: Molar Concentration in Mole per Liter (mol/L) Molar Concentration Unit Conversion
- Measurement: Mass Flux in Kilogram per Second per Square Meter (kg/s/m²)





Mass Flux Unit Conversion 🕑

- Measurement: **Density** in Kilogram per Cubic Meter (kg/m³) Density Unit Conversion
- Measurement: Diffusivity in Square Meter Per Second (m²/s)
 Diffusivity Unit Conversion
- Measurement: Molar Flux of Diffusing Component in Mole per Second Square Meter (mol/s*m²)
 Molar Flux of Diffusing Component Unit Conversion C
- Measurement: Time Inverse in 1 Per Second (1/s) Time Inverse Unit Conversion



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