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Important Formulas in Basics of Chemical Reaction Engineering & Forms of Reaction Rate

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List of 17 Important Formulas in Basics of Chemical Reaction Engineering & Forms of Reaction Rate

Important Formulas in Basics of Chemical Reaction Engineering & Forms of Reaction Rate

1) Feed Reactant Concentration

fx
$$C_{Ao} = \frac{F_{Ao}}{v_o}$$

Open Calculator 

ex
$$0.5\text{mol}/\text{m}^3 = \frac{5\text{mol}/\text{s}}{10\text{m}^3/\text{s}}$$

2) Number of Moles of Reactant Fed using Reactant Conversion

fx
$$N_{Ao} = \frac{N_A}{1 - X_A}$$

Open Calculator 

ex
$$30\text{mol} = \frac{9\text{mol}}{1 - 0.7}$$

3) Reactant Concentration of First Order Irreversible Reaction

fx
$$C = e^{-k' \cdot \Delta t} \cdot C_o$$

Open Calculator 

ex
$$20.99974\text{mol}/\text{m}^3 = e^{-2.508\text{s}^{-1} \cdot 0.5333\text{s}} \cdot 80\text{mol}/\text{m}^3$$



4) Reactant Concentration of Second Order Irreversible Reaction with Equal Reactant Conc using Time

$$\text{fx } C = \frac{1}{\left(\frac{1}{C_o}\right) + k'' \cdot \Delta t}$$

Open Calculator 

$$\text{ex } 22.2595\text{mol/m}^3 = \frac{1}{\left(\frac{1}{80\text{mol/m}^3}\right) + 0.0608\text{m}^3/(\text{mol}\cdot\text{s}) \cdot 0.5333\text{s}}$$

5) Reactant Concentration using Reactant Conversion

$$\text{fx } C = C_o \cdot (1 - X_A)$$

Open Calculator 

$$\text{ex } 24\text{mol/m}^3 = 80\text{mol/m}^3 \cdot (1 - 0.7)$$

6) Reactant Conversion using Molar Feed Rate of Reactant

$$\text{fx } X_A = 1 - \frac{F_A}{F_{A_o}}$$

Open Calculator 

$$\text{ex } 0.7 = 1 - \frac{1.5\text{mol/s}}{5\text{mol/s}}$$

7) Reactant Conversion using Number of Moles of Reactant Fed

$$\text{fx } X_A = 1 - \frac{N_A}{N_{A_o}}$$

Open Calculator 

$$\text{ex } 0.7 = 1 - \frac{9\text{mol}}{30\text{mol}}$$



8) Reactant Conversion using Reactant Concentration

$$\text{fx } X_A = 1 - \left(\frac{C}{C_o} \right)$$

Open Calculator 

$$\text{ex } 0.7 = 1 - \left(\frac{24\text{mol/m}^3}{80\text{mol/m}^3} \right)$$

9) Reacting Fluid Volume using Reaction Rate

$$\text{fx } V_{\text{fluid}} = \frac{\Delta n}{r \cdot \Delta t}$$

Open Calculator 

$$\text{ex } 2.500156\text{m}^3 = \frac{4\text{mol}}{3\text{mol/m}^3 \cdot \text{s} \cdot 0.5333\text{s}}$$

10) Reaction Rate based on Volume of Reacting Fluid

$$\text{fx } r = \frac{\Delta n}{V_{\text{fluid}} \cdot \Delta t}$$

Open Calculator 

$$\text{ex } 3.000188\text{mol/m}^3 \cdot \text{s} = \frac{4\text{mol}}{2.5\text{m}^3 \cdot 0.5333\text{s}}$$

11) Reaction Rate in Gas-Solid System

$$\text{fx } r = \frac{\Delta n}{V_{\text{solid}} \cdot \Delta t}$$

Open Calculator 

$$\text{ex } 2.988235\text{mol/m}^3 \cdot \text{s} = \frac{4\text{mol}}{2.51\text{m}^3 \cdot 0.5333\text{s}}$$



12) Reaction Rate in Reactor

$$\text{fx } r = \frac{\Delta n}{V_{\text{reactor}} \cdot \Delta t}$$

Open Calculator 

$$\text{ex } 3.012236 \text{mol/m}^3 \cdot \text{s} = \frac{4 \text{mol}}{2.49 \text{m}^3 \cdot 0.5333 \text{s}}$$

13) Reaction Time Interval of Gas-Solid System using Reaction Rate

$$\text{fx } \Delta t = \frac{\Delta n}{r \cdot V_{\text{solid}}}$$

Open Calculator 

$$\text{ex } 0.531208 \text{s} = \frac{4 \text{mol}}{3 \text{mol/m}^3 \cdot \text{s} \cdot 2.51 \text{m}^3}$$

14) Reaction Time Interval of Reacting Fluid using Reaction Rate

$$\text{fx } \Delta t = \frac{\Delta n}{r \cdot V_{\text{fluid}}}$$

Open Calculator 

$$\text{ex } 0.533333 \text{s} = \frac{4 \text{mol}}{3 \text{mol/m}^3 \cdot \text{s} \cdot 2.5 \text{m}^3}$$

15) Reaction Time Interval of Reactor using Reaction Rate

$$\text{fx } \Delta t = \frac{\Delta n}{r \cdot V_{\text{reactor}}}$$

Open Calculator 

$$\text{ex } 0.535475 \text{s} = \frac{4 \text{mol}}{3 \text{mol/m}^3 \cdot \text{s} \cdot 2.49 \text{m}^3}$$



16) Reactor Volume using Reaction Rate

[Open Calculator !\[\]\(feabb98897b440bc8695a03336a6e2df_img.jpg\)](#)

fx
$$V_{\text{reactor}} = \frac{\Delta n}{r \cdot \Delta t}$$

ex
$$2.500156\text{m}^3 = \frac{4\text{mol}}{3\text{mol}/\text{m}^3 \cdot \text{s} \cdot 0.5333\text{s}}$$

17) Solid Volume using Reaction Rate

[Open Calculator !\[\]\(642aa997563f9a325b310230bb5078b7_img.jpg\)](#)

fx
$$V_{\text{solid}} = \frac{\Delta n}{r \cdot \Delta t}$$

ex
$$2.500156\text{m}^3 = \frac{4\text{mol}}{3\text{mol}/\text{m}^3 \cdot \text{s} \cdot 0.5333\text{s}}$$












Variables Used

- **C** Reactant Concentration (Mole per Cubic Meter)
- **C_{A0}** Concentration of Key Reactant A in the Feed (Mole per Cubic Meter)
- **C₀** Initial Reactant Concentration (Mole per Cubic Meter)
- **F_A** Molar Flow Rate of Unreacted Reactant (Mole per Second)
- **F_{A0}** Molar Feed Rate of Reactant (Mole per Second)
- **k'** Rate Constant for First Order Reaction (1 Per Second)
- **k''** Rate Constant for Second Order Reaction (Cubic Meter per Mole Second)
- **N_A** Number of Moles of Unreacted Reactant-A (Mole)
- **N_{A0}** Number of Moles of Reactant-A Fed (Mole)
- **r** Reaction Rate (Mole per Cubic Meter Second)
- **V_{fluid}** Fluid Volume (Cubic Meter)
- **v₀** Volumetric Flow Rate of Feed to Reactor (Cubic Meter per Second)
- **V_{reactor}** Reactor Volume (Cubic Meter)
- **V_{solid}** Solid Volume (Cubic Meter)
- **X_A** Reactant Conversion
- **Δn** Change in Number of Moles (Mole)
- **Δt** Time Interval (Second)














Constants, Functions, Measurements used

- **Constant:** **e**, 2.71828182845904523536028747135266249
Napier's constant
- **Measurement:** **Time** in Second (s)
Time Unit Conversion 
- **Measurement:** **Amount of Substance** in Mole (mol)
Amount of Substance Unit Conversion 
- **Measurement:** **Volume** in Cubic Meter (m^3)
Volume Unit Conversion 
- **Measurement:** **Volumetric Flow Rate** in Cubic Meter per Second (m^3/s)
Volumetric Flow Rate Unit Conversion 
- **Measurement:** **Molar Flow Rate** in Mole per Second (mol/s)
Molar Flow Rate Unit Conversion 
- **Measurement:** **Molar Concentration** in Mole per Cubic Meter (mol/m^3)
Molar Concentration Unit Conversion 
- **Measurement:** **Reaction Rate** in Mole per Cubic Meter Second ($\text{mol/m}^3\cdot\text{s}$)
Reaction Rate Unit Conversion 
- **Measurement:** **First Order Reaction Rate Constant** in 1 Per Second (s^{-1})
First Order Reaction Rate Constant Unit Conversion 
- **Measurement:** **Second Order Reaction Rate Constant** in Cubic Meter per Mole Second ($\text{m}^3/(\text{mol}\cdot\text{s})$)
Second Order Reaction Rate Constant Unit Conversion 



Check other formula lists

- [Basics of Chemical Reaction Engineering Formulas](#) 
- [Basics of Parallel & Single Reactions Formulas](#) 
- [Basics of Reactor Design and Temperature Dependency from Arrhenius Law Formulas](#) 
- [Forms of Reaction Rate Formulas](#) 
- [Important Formulas in Basics of Chemical Reaction Engineering & Forms of Reaction Rate](#) 
- [Important Formulas in Constant and Variable Volume Batch Reactor](#) 
- [Important Formulas in Constant Volume Batch Reactor for First, Second & Third Order Reaction](#) 
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