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Important Formulas of Basic Chemistry

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List of 9 Important Formulas of Basic Chemistry

Important Formulas of Basic Chemistry ↗

1) Boiling Point ↗

fx $bp = bp_{\text{solvent}} + \Delta bp$

[Open Calculator ↗](#)

ex $961.2\text{K} = 80.1\text{K} + 12\text{K}$

2) Bond Order ↗

fx $B.O = \left(\frac{1}{2}\right) \cdot (B e^- - A.B e^-)$

[Open Calculator ↗](#)

ex $2 = \left(\frac{1}{2}\right) \cdot (8 - 4)$

3) Change in Boiling Point of Solvent ↗

fx $\Delta bp = K_b \cdot m$

[Open Calculator ↗](#)

ex $12\text{K} = 4.8 \cdot 2.5\text{mol/L}$



4) Molar Volume ↗

$$fx \quad v_m = \frac{A \cdot M_{molar}}{\rho}$$

Open Calculator ↗

$$ex \quad 1.2E^{-6} m^3/mol = \frac{28.085g \cdot 44.01g/mol}{997kg/m^3}$$

5) Mole Fraction ↗

$$fx \quad X = \frac{n}{n + N}$$

Open Calculator ↗

$$ex \quad 0.398726 = \frac{3.4483mol}{3.4483mol + 5.2mol}$$

6) Molecular Formula ↗

$$fx \quad M.F = \frac{M_{molar}}{EFM}$$

Open Calculator ↗

$$ex \quad 2442.286 = \frac{44.01g/mol}{0.01802g}$$

7) Partition Coefficient ↗

$$fx \quad K = \frac{CS}{cm}$$

Open Calculator ↗

$$ex \quad 1.0875 = \frac{0.087mol/L}{0.080mol/L}$$



8) Percent by Weight 

fx % by wt. = $\frac{\text{gSolute}}{100\text{gSolution}}$

Open Calculator 

ex $0.2 = \frac{20\text{g}}{100\text{g}}$

9) Specific Heat Capacity 

fx $c = \frac{Q}{M \cdot \Delta T_{\text{rise}}}$

Open Calculator 

ex $7.404795\text{kJ/kg}^*\text{K} = \frac{4200\text{J}}{35.45\text{g} \cdot 16\text{K}}$



Variables Used

- **% by wt.** Percent By Weight
- **100gSolution** 100 g of Solution (Gram)
- **A** Atomic Weight (Gram)
- **A.B e⁻** Number of Antibonding Electrons
- **B e⁻** Number of Bonding Electrons
- **B.O** Bond Order
- **bp** Boiling Point (Kelvin)
- **bp_{solvent}** Boiling Point of Solvent (Kelvin)
- **c** Specific Heat Capacity (Kilojoule per Kilogram per K)
- **cm** Concentration of Solute in Mobile Phase (Mole per Liter)
- **cs** Concentration of Solute in Stationary Phase (Mole per Liter)
- **EFM** Mass of Empirical Formulas (Gram)
- **gSolute** Gram of Solute (Gram)
- **K** Partition Coefficient
- **K_b** Molal Boiling Point Elevation Constant
- **m** Molal Concentration of Solute (Mole per Liter)
- **M** Mass (Gram)
- **M_{molar}** Molar Mass (Gram Per Mole)
- **M.F** Molecular Formula
- **n** Number of Moles of Solute (Mole)
- **N** Number of Moles of Solvent (Mole)
- **Q** Heat Energy (Joule)
- **V_m** Molar Volume (Cubic Meter per Mole)



- **X** Mole Fraction
- **Δbp** Change in Boiling Point of Solvent (*Kelvin*)
- **ΔT_{rise}** Rise in Temperature (*Kelvin*)
- **ρ** Density (*Kilogram per Cubic Meter*)



Constants, Functions, Measurements used

- **Measurement:** **Weight** in Gram (g)
Weight Unit Conversion 
- **Measurement:** **Temperature** in Kelvin (K)
Temperature Unit Conversion 
- **Measurement:** **Amount of Substance** in Mole (mol)
Amount of Substance Unit Conversion 
- **Measurement:** **Energy** in Joule (J)
Energy Unit Conversion 
- **Measurement:** **Specific Heat Capacity** in Kilojoule per Kilogram per K (kJ/kg*K)
Specific Heat Capacity Unit Conversion 
- **Measurement:** **Molar Concentration** in Mole per Liter (mol/L)
Molar Concentration Unit Conversion 
- **Measurement:** **Density** in Kilogram per Cubic Meter (kg/m³)
Density Unit Conversion 
- **Measurement:** **Molar Mass** in Gram Per Mole (g/mol)
Molar Mass Unit Conversion 
- **Measurement:** **Molar Magnetic Susceptibility** in Cubic Meter per Mole (m³/mol)
Molar Magnetic Susceptibility Unit Conversion 



Check other formula lists

- [Determination of Equivalent Mass](#) ↗
- [Important Formulas of Basic Chemistry](#) ↗

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